

Polyacrylonitrile Nanofiber Optimization as Precursor of Carbon Nanofibers for Supercapacitors

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Abstract

Polyacrylonitrile (PAN) nanofibers are one of the primary precursors in the production of carbon nanofibers. The nanofiber morphology is significantly affected by the process parameters such as polymer concentration, distance, applied voltage and feed rate during the production of PAN nanofibers obtained by the solution-based electrospinning method, and these parameters should be optimized properly. In this study, firstly PAN nanofiber production parameters were optimized, and then homogeneous and thin PAN nanofibers produced in optimum conditions were used as the precursor in the production of carbon nanofibers. PAN nanofibers with a diameter of 233 nm were obtained at 7.5% PAN concentration in N,N-dimethylformamide (DMF), 28 kV applied voltage, 17.5 cm nozzle to collector distance, 2 ml/h feed rate and 500 rpm rotation speed of the aluminum drum. The carbon nanofiber diameters produced after the stabilization and carbonization processes were measured as 200 and 140 nm, respectively. The morphological, chemical and thermal properties of the produced nanofibers were characterized by field emission scanning electron microscopy (FE-SEM), Fourier transform infrared (FT-IR), thermogravimetric analyzer (TGA). Carbon nanofibers, which are made from optimized electrospun PAN nanofibers, can be used to construct supercapacitors in future studies.

Keywords: Carbonization, Carbon Nanofiber, Electrospinning, PAN Nanofiber, Supercapacitor.

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1. Introduction

Carbon nanofibers are nanomaterials with nanoscale diameter that have been used in many application areas such as supercapacitors, rechargeable batteries, dye-sensitized solar cells, catalysis, sensors, and adsorption-desorption membranes as well as in the biomedical field. Therefore, they have attracted great attention in recent years [1–8]. Although carbon nanofibers can be produced via vapor growth or plasma-enhanced chemical vapor deposition techniques, the most common production is achieved through the stabilization and carbonization of polyacrylonitrile (PAN) nanofibers precursor due to the low cost, continuous, simple and easily controllable production process [3,9,10]. On the other hand, electrospinning is the most widely used method in nanofiber production. Electrospinning is generally divided into two types: (a) melt electrospinning and (b) solution-based electrospinning. The solution-based electrospinning method stands out because of the fact that nanofibers can be produced from various polymers such as polyacrylonitrile (PAN), poly (vinylidene fluoride) (PVDF), poly(vinyl alcohol) (PVA), polyamide 6,6, polycaprolactone (PCL), poly (vinylidene fluoride-co-hexafluoropropylene) (PVDF-co-HFP), thermoplastic polyurethane (TPU) and that the nanofiber properties can be controlled more easily [11–19]. The parameters affecting nanofiber morphology in the solution-based electrospinning method can be listed as temperature, humidity, solution viscosity (polymer concentration), the distance between nozzle and collector electrode, feed rate and the applied potential [20–22]. The determination of optimum process parameters is important to obtain nanofibers with desired properties.

This study aimed to investigate the influence of polymer concentration, distance, feed rate and applied voltage on the morphology of PAN nanofibers. The PAN nanofiber production was carried out using the production parameters determined by the production rate, continuity and nanofiber morphology. After the PAN nanofibers were stabilized in an air environment at 90 °C for 90 minutes, the stabilized nanofibers were successfully carbonized at 1000 °C for 3 hours in an argon atmosphere, and carbon nanofibers were successfully obtained. The current study also tried to investigate the chemical, thermal and morphological properties of the obtained PAN, stabilized PAN (St-PAN) and carbon nanofibers.

2. Material and Methods

2.1. Materials

Polyacrylonitrile (PAN) (M_w =150000 g/mole) was obtained from J&K Scientific (Shanghai, China), N,N-dimethylformamide (DMF, \geq 99wt%) was purchased from Sigma-Aldrich and used as received without further purification.

2.2. Optimization of PAN Nanofiber Production

2.2.1. Preparation of Polymer Solutions

The required amount of PAN was added into DMF and stirred overnight at 80 °C. PAN solutions of 7.5%, 10%, 12.5%, 15% by mass were used in the study.

2.2.2. Optimization of Electrospinning Parameters

Inovenso Nanospinner 24 electrospinning device was used to produce nanofiber mats. Figure 1 illustrates the electrospinning setup. In this study, the effects of 4 different parameters on nanofiber production were investigated. For this purpose, nanofibers were produced using 4 different polymer concentrations (7.5, 10, 12.5, 15 wt.%), 4 different nozzle-to-collector distance (12.5, 15, 17.5, 20 cm), 4 different feed rates (1, 1.5, 2, 3 ml/h), 3 different applied voltages (25, 30, 35 kV). The electrospinning conditions listed in Table 1 were used the produce nanofiber mats. According to the parameters applied here, the sample codes were named S1-S13, which are presented in Table 1 with electrospinning conditions.



Figure 1. Schematic illustration of electrospinning process.

	Sample Code	Variable Parameter	Constant Parameters		
Concentration Effect		Concentration (wt.%)	Distance (cm)	Feed Rate (ml/h)	Voltage (V)
	S1	7.5%	20 cm	2 ml/h	30 kV
	S2	10%	20 cm	2 ml/h	30 kV
	S 3	12.5%	20 cm	2 ml/h	30 kV
	S4	15%	20 cm	2 ml/h	30 kV
Distance Effect		Distance (cm)	Concentration (wt.%)	Feed Rate (ml/h)	Voltage (V)
	S5	12.5 cm	12.5%	2 ml/h	30 kV
	S 6	15 cm	12.5%	2 ml/h	30 kV
	S7	17.5 cm	12.5%	2 ml/h	30 kV
	S 3	20 cm	12.5%	2 ml/h	30 kV
Feed Rate Effect		Feed Rate (ml/h)	Distance (cm)	Concentration (wt.%)	Voltage (V)
	S8	1 ml/h	15 cm	12.5%	30 kV
	S9	1.5 ml/h	15 cm	12.5%	30 kV
	S6	2 ml/h	15 cm	12.5%	30 kV
	S10	3 ml/h	15 cm	12.5%	30 kV
Applied Voltage Effect		Voltage (V)	Distance (cm)	Concentration (wt.%)	Feed Rate (ml/h)
	S11	25 kV	20 cm	15%	2 ml/h
	S4	30 kV	20 cm	15%	2 ml/h
	S12	35 kV	20 cm	15%	2 ml/h
Optimum Sample	S13	28 kV	17.5 cm	7.5%	2 ml/h

Table 1. Electrospinning Parameters of PAN Nanofiber Production

2.3. Production of Carbon Nanofibers

Among the conditions specified above, 7.5 wt.% PAN in DMF, 28 kV applied voltage, 17.5 cm nozzle-to-collector distance and 2 ml/h feed rate were determined as optimum electrospinning conditions, and PAN nanofibers mats were produced. For the stabilization of PAN nanofibers, it was heated up to 300 °C with a heating rate of 2 °C/min and kept at 300 °C for 90 min in the air atmosphere. The stabilized PAN nanofibers were carbonized by gradual heating in the tube furnace in an argon atmosphere. First, it was heated up to 280 °C with a heating rate of 10 °C/min and waited for 1 hour, then it was heated up to 1000 °C with a heating rate of 5 °C/min and kept for 3 hours at 1000 °C. At the end of the process, carbon nanofibers were successfully obtained. Figure 2 demonstrates the time-temperature profile during stabilization and carbonization.



Figure 2. Time-temperature profile during a) Stabilization, b) Carbonization processes.

2.4. Characterization

The chemical structure of nanofibers was characterized by Fourier transform infrared spectroscopy (FT-IR) in the range of 4000-400 cm⁻¹ with a 4 cm⁻¹ spectral resolution on a Thermo Scientific Nicolet i550 model FTIR spectrometer with Smart Orbit-Diamond model ATR on transmittance mode. The morphology of nanofibers was investigated via field emission scanning electron microscope (FE-SEM) (Zeiss-Gemini 300) with field emission electron gun set at 20 kV and 10 mm working distance. The diameter of nanofiber was calculated via ImageJ 1.51j8 software based on SEM images. Thermogravimetric analysis (TGA) was carried out in the temperature range from 30 to 900 °C under a nitrogen atmosphere at a rate of 10 °C min⁻¹ on a Perkin Elmer– STA6000 (Perkin Elmer, Waltham, MA, USA) thermogravimetric analyzer.

3. Results and Discussion

One of the most important parameters affecting the nanofiber morphology is the viscosity of the polymer solution. The solution viscosity is also directly proportional to the polymer concentration and the molecular weight of the polymer. The studies in the relevant literature reveals that uniform nanofibres cannot be obtained and beads are observed as a result of the solution viscosity below a certain value [23–27]. Moreover, too high viscosity adversely affects fiber formation and increases nanofiber diameters.

Figure 3 shows the SEM images of PAN nanofibers obtained from PAN solutions at different polymer concentrations (7.5 wt%, 10 wt%, 12.5 wt%, 15 wt%). While S1, S2 and S3 had more homogeneous and uniform nanofiber surfaces, it was seen in the S4 sample that fiber thicknesses increase, and joints occur at the intersection points of nanofibers. The average diameters of nanofibers are shown in Figure 5-a. The diameters of S1, S2, S3 and S4 samples were measured as 275 ± 57 , 334 ± 67 , 575 ± 101 , 1925 ± 401 nm, respectively. The increase in polymer concentration in the solution appears to enhance viscosity; consequently, this has a huge effect on the nanofiber diameter.



Figure 3. SEM image of nanofibers produced at different polymer concentrations **a**) S1 (10k x), **b**) S2 (10k x), **c**) S3 (10k x), **d**) S4 (1k x).

Figure 4 demonstrates the SEM images of nanofibers produced at the different nozzle to collector distance. It was observed that all nanofibers obtained are smooth and homogeneous. Figure 5-b displays the average diameter of nanofibers produced at the different distances. S5, S6, S7, S3 diameters were measured as 549 ± 113 , 508 ± 99 , 526 ± 72 , 575 ± 101 nm, respectively.

Although some studies in the related literature mention that the increase in distance decreases the diameter of the nanofiber, no direct correlation has been observed in our study [28,29]. Considering the standard deviations, a decreasing trend of 12.5 cm to 17.5 cm appears, but the nanofibers produced at a distance of 20 cm are the thickest nanofibers. It is thought that the effect caused by parameters such as temperature and humidity that we cannot control due to being produced at different times may have suppressed the effect caused by nozzle to collector distance.



Figure 4. SEM image of nanofibers produced at different nozzle-to-collector distance a) S5 (10k x), b) S6 (10k x), c) S7 (10k x), d) S3 (10k x).



Figure 5. a) Polymer concentration and b) Nozzle to collector distance effect on NF diameter.

Figure 6 illustrates the SEM images of nanofibers at different feed rates while the nanofiber diameters are plotted in Figure 8-a. Nanofiber diameters were calculated as 560 ± 151 , 587 ± 153 , 508 ± 99 , 496 ± 99 nm, respectively. Most studies in the relevant literature suggest that the increase in feed rate increases the nanofiber diameter [30–33]. However, based on our experience on nanofiber, we can say that this effect is not seen in every feed rate increase. This increase is directly related to the nozzle diameter and the device configuration. No significant relationship has been observed in the above mentioned feed rate differences with the device we use. It is possible to observe this diameter increase in much higher feed rate differences. On the other hand, some studies state that the increase in feed rate decreases the nanofiber diameter [21].



Figure 6. SEM image of nanofibers produced at different feed rates a) S8 (10k x), b) S9 (10k x), c) S6 (10k x), d) S10 (10k x).

Another important parameter of the electrospinning process is the applied voltage. It is mentioned in the literature that the applied voltage affects the initial droplet formation, feed rate and viscosity [34]. The electric field applied to the polymer droplet overcomes the surface tension of the droplet and creates a jet [35]. Nanofiber morphology can be manipulated by controlling the applied voltage. Increasing the applied voltage generally causes the nanofiber diameter to decrease [20,34,36,37]. However, when the critical voltage value is exceeded, an increase in fiber diameter can be seen due to the decrease in flight time. In addition, increasing the applied voltage causes multiple jet formation during production [38,39].

Figure 7 shows the SEM images of nanofibers produced at different applied voltage values. At low voltage values, nanofibers appeared to stick together and move away from the fiber appearance. It was seen that more uniform and homogeneous nanofibers are obtained with an increase in voltage. Also in Figure 8-b, the diameter values of S11, S4 and S12 were measured as 1937±548, 1925±401 and 1917±339 nm, respectively. As a result, it was seen that thinner and neat nanofibers are obtained by increasing the voltage.



Figure 7. SEM image of nanofibers produced at different applied voltages a) S11 (2k x), b) S4 (2k x), c) S12 (2k x).



Figure 8. a) Feed rate b) Applied voltage effect on NF diameters.

Optimum parameters were determined as 7.5 wt% PAN concentration, 28 kV applied voltage, 2 ml/hour feed rate and 17.5 cm nozzle to collector distance based on the studies above mentioned considering production speed and production continuity. PAN nanofibers prepared at these parameters were produced, and then they were stabilized and carbonized. Figure 9 displays PAN nanofiber, stabilized PAN nanofiber and carbon nanofiber.



Figure 9. a) PAN NF (S13), b) Stabilized PAN NF, c) Carbon NF.

Figure 10 shows the average fiber diameters of PAN NF, Stb-PAN NF and carbon nanofibers, and Figure 11 displays SEM images of the samples. The average diameters of PAN NF were measured as 233 ± 32 nm, it dropped to 209 ± 16 nm after the stabilization process. After carbonization, the diameter of carbon nanofibers was calculated as 140 ± 17 nm. As seen in Figure 11-c, homogeneous and thin carbon nanofibers were successfully obtained.



Figure 10. Diameter of PAN nanofiber, stabilized PAN nanofiber and carbon nanofiber.



Figure 11. SEM images of a) PAN nanofiber, b) stabilized PAN nanofiber and c) carbon nanofiber.

Figure 12 illustrates the FT-IR spectra of PAN nanofiber, St-PAN nanofiber and carbon nanofiber. The characteristic absorption band of PAN nanofiber due to -C=N stretching can be seen in 2242 cm⁻¹. Vibration of aliphatic CH groups (CH, CH₂, and CH₃) give peaks at 2939 cm⁻¹, 1453 cm⁻¹, 1357 cm⁻¹, and 1251 cm⁻¹ [40,41].

During stabilization, C=N bonds in the PAN structure were broken and C=N bonds were formed. This caused the intensity of C=N peaks seen at 2242 cm⁻¹ to decrease and shift to 2230 cm⁻¹. In addition, new peaks were formed as a result of the overlapping of the vibration peaks at different frequencies of functional groups such as C=C, C=N, C=O, and N-H between 1760-980 cm⁻¹ [42-44]. The peak was observed at 803 cm⁻¹ due to the vibration of aromatic C-H bonds formed due to oxidative dehydrogenation in the presence of oxygen [44,45]. Within the scope of the studies in the relevant literature, the stabilization process of PAN was carried out in the range of 210-300 °C [17,42-44,46,47]. As the stabilization temperature increased, the oxidation rate increased. For this reason, 300 °C was chosen as the stabilization process carried out at 1000 °C. These results indicated that the stabilization and carbonization processes were successfully performed.



Figure 12. FT-IR spectrum of PAN NF, St-PAN NF, Carbon NF.

Figure 13 demonstrates TGA thermogram of PAN NF and St-PAN NF. PAN NF primarily lost its moisture at around 100 °C. Afterwards, it was decomposed between 300-500 °C and left 29% residue around 900 °C. During this degradation, random chain scission, cyclization, dehydrogenation, and cross-linking took place [48]. On the other hand, St-PAN NF sample started to carbonize around 300 °C after removing the moisture up to 100 °C [17]. The stabilization of PAN nanofiber increased carbon yield at the end of the carbonization process.



Figure 13. TGA thermogram of PAN NF and St-PAN NF.

4. Conclusion

In this study, PAN polymer was electrospun on an aluminum drum rotating at 500 rpm at different polymer concentrations, applied voltages, feed rates, and nozzle to collector distances, and it aimed to investigate the effect of these parameters on PAN nanofiber morphology. In the light of study results, it was observed that the most important parameter affecting polymer morphology was polymer concentration, and the nanofiber diameter decreased with decreasing polymer concentration. Although the studies in the related literature stated that increasing the distance decreases the nanofiber diameter, no correlative relationship was observed between the distances examined in this study. In addition, unlike the most studies in the literature, a decrease in nanofiber diameter was observed with increasing feed rates generally. It was perceived that more smooth and thin nanofibers were obtained with the increase of applied voltage. Taking the parameters tested into account, polymer solution concentration, the distance between nozzle and collector electrode, feed rate, and the applied potential were determined as 7.5 w%, 17.5 cm, 2 ml/h, and 28 kV, respectively; and then PAN nanofibers were produced. The diameters of PAN nanofibers produced under these conditions were measured as 233±32 nm and used as a precursor for carbon nanofiber production. The PAN nanofiber precursor was first stabilized in the air atmosphere at 280 °C, and then it was carbonized in argon atmosphere at 1000 °C to obtain carbon nanofibers. After stabilization and carbonization, nanofiber diameters were calculated as 209±16 nm and 140±17 nm, respectively. FT-IR analysis confirmed that the stabilization and carbonization processes have been successfully performed. On the other hand, TGA results showed that the thermal stability of stabilized PAN nanofibers is higher than that of PAN nanofibers. This optimization study is of utmost importance since it produces carbon nanofiber material under the optimum conditions and it provides guidance for future supercapacitor applications.

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